

Advancing Coal Catalytic Gasification to Promote Optimum Syngas Production

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Proposed Work

- Novel approach will be the use of red mud catalysts to improve the gasification process.
- Adding small fractions of biomass feedstock will help increase hydrogen content for quality syngas suitable for hydrocarbon synthesis.
- The proposed research will focus on:
 - bench-scale and pilot-scale experiments to measure key reactive properties,
 - developing kinetic models that predict the product formation, and
 - CFD modeling that incorporates the reaction kinetics for catalytic coal gasification.
- Collaboration consists of a unique team with expertise in experiments, modeling reaction kinetics, combustion and CFD of multiphase flows.

Specific Objectives

- Measure syngas composition
 - Perform experiments to determine gas composition of sub-bituminous coal with red mud catalysts and modified with nickel
 - Investigate catalytic gasification to reduce methane
 - Add small quantities of biomass to increase hydrogen content of syngas
- Model detailed reaction kinetics and product formation
 - Analyze reactions from catalytic gasification to understand major pathways involved
 - Identify necessary reaction mechanisms needed for the computational models
 - Implement reaction kinetics into MFiX CFD code.
 - Validate with the experiments

Summary

- Red mud promoted better cracking of char and tars into gas product compared to silica sand.
- Product gas yields obtained using red mud were higher than product gas yields from silica sand experiments.
- Gasification of coal improved with increasing temperature and addition of biomass feedstock.
- Gasification experiments provided data to develop a chemical kinetics model, and reaction rates were tuned for implementation into CFD.
- Coal gasification simulations using an earlier version of the lumped kinetic model demonstrated the need for parameter tuning to validated well with experiments.
- C3M-HPTR pyrolysis model validates well at 900°C but yields over-prediction of CO at lower temperatures.

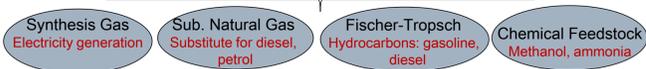
Final kinetics models will be tuned according to co-gasification experiments and CFD input

Gasification of Coal/Biomass Mixtures on RM

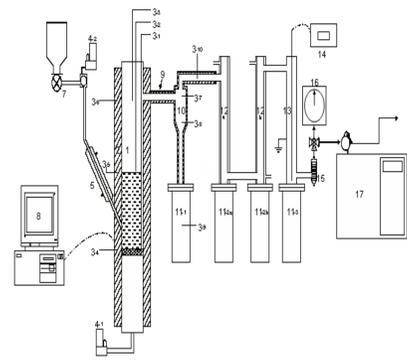


Sub-bituminous Coal, CO₂, NO_x and SO_x

- Coal gasification is an alternative to mitigate pollutant emissions
- Blending coal with biomass allows to control the H₂ to CO ratio



Bench-scale Fluidized Bed Reactor



- Fluidized bed reactor
- Thermocouple
- Mass flow controller
- Jacketed air-cooled feeder tube
- Hopper
- Screw feeder
- Computer
- Heating tape
- Hot gas filter
- Reservoir
- Condenser
- ESP
- AC power supply
- Filter
- Wet gas meter
- Gas chromatograph

Coal Gasification under Steam on RM

Product gases (mol, %) for coal gasification on RM under N₂ and under N₂ & Steam

Temp, °C	H ₂ : CO	H ₂	CH ₄	CO	CO ₂	C ₂ -C ₄	O ₂	N ₂	Σ _{Gases}
700	3.5 : 1	5.032	0.809	1.442	1.988	0.480	0.059	89.690	99.50
800	2.5 : 1	5.939	0.862	2.373	2.933	0.121	0.036	86.521	98.79
900	1.6 : 1	8.264	0.938	5.161	3.010	0.367	0.040	80.052	97.83

Product liquids, solids, and gases (wt, %)

Temp, °C	Coal on RM under N ₂			Coal on RM under N ₂ & Steam		
	Moist., %	Liquid	Solid	Moist., %	Liquid	Solid
700	24.19	29.58	43.20	23.96	19.36	54.06
800	24.33	27.57	42.78	22.52	17.08	40.81
900	24.94	20.97	29.61	21.55	13.94	31.87

Coal Gasification under CO₂ and Steam on RM

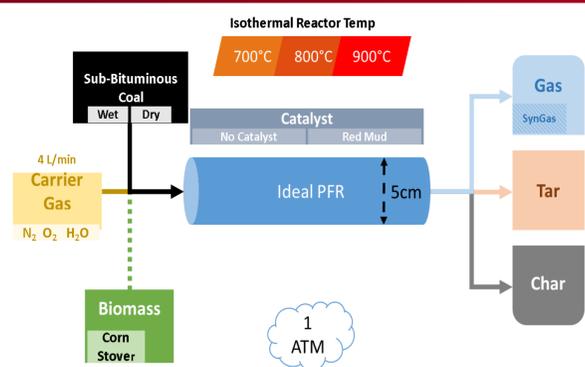
Product gases (mol, %) for wet coal gasification under N₂ and CO₂ on RM

Temp, °C	H ₂ : CO	H ₂	CH ₄	CO
800	2.5 : 1	5.94	0.86	2.37
900	1.6 : 1	8.26	0.94	5.16

Product liquids, solids, and gases (wt, %)

Temp, °C	Wet Coal on RM under N ₂			Wet Coal under CO ₂ on RM		
	Moisture, %	Liquid	Solid	Moisture, %	Liquid	Solid
800	24.33	27.57	42.78	24.11	37.03	43.56
900	24.94	20.97	29.61	24.11	37.03	43.56

Modeling the Kinetics of Product Formation



Kinetics Strategy and Reaction Network

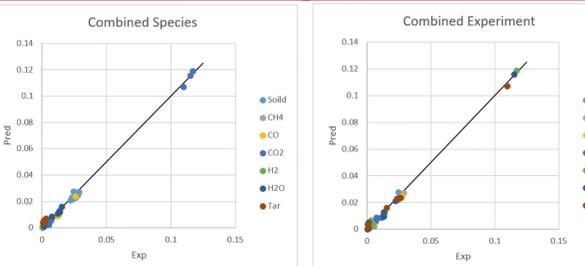
Lump	MW (g/mol)	Density (g/L)	OBJ Lump	OBJ Weight a
Coal	1000	1300		
CharH (C2H)	25	2000	Solid	1
CharC (C)	12	2000		
Tar	182	---	Tar	100
H ₂ O	18	---	H ₂ O	1
CH ₄	16	---	CH ₄	1
CO	28	---	CO	0.1
CO ₂	44	---	CO ₂	1
H ₂	2	---	H ₂	0.1

$$OBJ = \sum_i^n \left(\frac{exp_i - mdl_i}{exp_i * \alpha} \right)^2 = 1.47$$

Reaction	Type	logA	E*
Coal → a CharH + b CH ₄ + c CO + d CO ₂ + e H ₂ + f H ₂ O	Pyrolysis	4.70	24.20
Coal → g Tar + h CO ₂ + i H ₂ O	Pyrolysis	-0.37	1.69
Tar → j CharC + k CH ₄ + l H ₂ + m H ₂ O	Condensation	2.22	11.39
CharH → 2 CharC + 1/2 H ₂	Pyrolysis	4.77	29.48
CharC + H ₂ O → CO + H ₂	Gasification	-0.22	3.37
CharC + CO ₂ → 2 CO	Gasification	-0.002	0.030
CharC + 2 H ₂ → CH ₄	Gasification	-6.77	9.93
CO + H ₂ O ↔ CO ₂ + H ₂	Water-Gas-Shift	2.45	1.22

a	25.8	e	10.8	i	4.4	m	1.1
b	1	f	0.11	j	11.5		
c	8.2	g	4.7	k	1.5		
d	1.9	h	1.3	l	0.45		

Kinetics Model Results

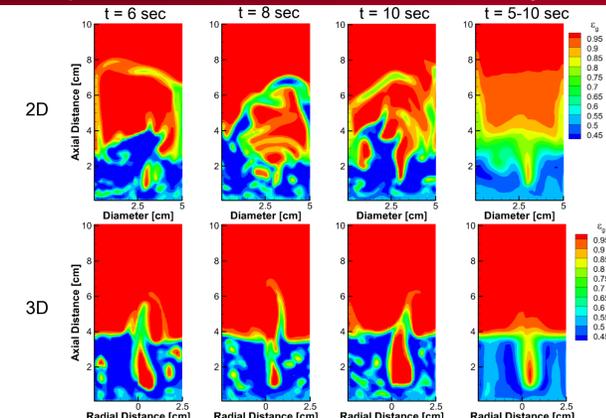


	Sim_0	Sim_1	Sim_2	Sim_3	Sim_4	Sim_6	Sim_6
Temp °C	900	700	800	900	700	800	900
Carrier Gas	N ₂	N ₂	N ₂	N ₂	CO ₂	CO ₂	CO ₂
Fuel	Dry Coal	Wet Coal	Wet Coal	Wet Coal	Wet Coal	Wet Coal	Wet Coal

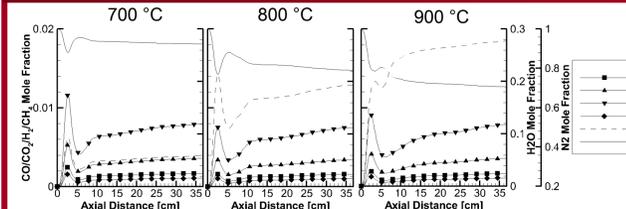
Computational Fluid Dynamics Simulations

- Simulations were performed in MFiX (2014-1) using the Two-Fluid (Eulerian-Eulerian) Model
- Coal is introduced directly into the sand bed using a point source strategy where experimental carrier stream velocities and mass flow rates are preserved
- Preliminary simulations were performed under an N₂ atmosphere implementing the lumped drying and pyrolysis reactions to assist in kinetics model tuning
- C3M-HPTR chemistry model was also tested because of fuel similarity

Comparison of 2D and 3D Point Source Injection



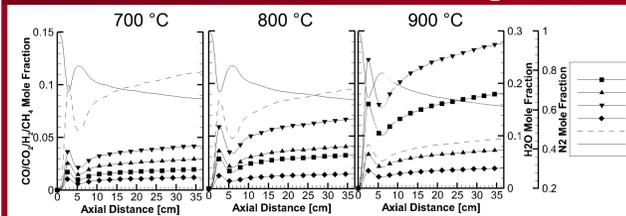
Simulation Results for Preliminary Lumped Model



Comparison of N₂/H₂O/Tar Free Gas Phase Mole Fractions and Relative Error

	700 °C			800 °C			900 °C		
	Exp.	Sim.	% RE	Exp.	Sim.	% RE	Exp.	Sim.	% RE
CO	0.190	0.118	38.2	0.209	0.118	43.8	0.318	0.118	63.8
CO ₂	0.286	0.252	11.8	0.265	0.252	5.0	0.128	0.252	96.2
H ₂	0.407	0.555	36.8	0.434	0.555	27.9	0.485	0.555	14.3
CH ₄	0.118	0.076	36.1	0.092	0.078	17.6	0.068	0.078	11.1

Simulation Results for C3M-HPTR N₂ Pyrolysis



Comparison of N₂/H₂O/Tar Free Gas Phase Mole Fractions and Relative Error

	700 °C			800 °C			900 °C		
	Exp.	Sim.	% RE	Exp.	Sim.	% RE	Exp.	Sim.	% RE
CO	0.190	0.323	70.0	0.209	0.310	48.2	0.318	0.303	4.8
CO ₂	0.286	0.239	16.4	0.265	0.172	35.2	0.128	0.137	6.2
H ₂	0.407	0.329	18.8	0.434	0.417	3.9	0.485	0.463	4.7
CH ₄	0.118	0.108	8.6	0.092	0.102	10.5	0.068	0.098	43.8